

Basic Stoichiometry PhET Lab rvsd 2/2011

Let's make some sandwiches!

Introduction:

When we bake/cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar. YUCK! In chemistry, reactions proceed with very specific recipes. The study of these recipes is *stoichiometry*. When the reactants are present in the correct amounts, the reaction will produce products. What happens if there are more or less of some of the reactants present?

Procedure: Go to: <u>http://phet.colorado.edu/en/simulation/reactants-products-and-leftovers</u> Click the play button

Part 1: Making Sandwiches:

Cheese Sandwiches

- 1. The Cheese Sandwhich is a simulation of a two-reactant *synthesis* reaction. In this case, one reactant will be *limiting*, while the other will be in excess.
- 2. Take some time and familiarize yourself with the simulation.
- 3. Set the reaction to a simple mole ratio of 2:1
- 4. Complete the table below while making tasty cheese sandwiches (Enter values into reactants and products section at bottom for simultation)

Bread Used	Cheese Used	Sandwiches Made	Excess Bread	Excess Cheese
5 slices	5 slices			
4 slices	3 slices			
		2 sandwiches	1 slice	0 slices
6 slices		3 sandwiches		4 slices

Meat and Cheese Sandwiches

- 1. The Meat and Cheese Sandwhich is a simulation of a three-reactant *synthesis* reaction. In this case, one reactant will be *limiting*, while the other two will be in excess.
- 2. Take some time and familiarize yourself with the simulation.
- 3. Set the reaction to a simple mole ratio of 2:1:1
- 4. Complete the table below while making tasty cheese sandwiches (Enter values into reactants and products section at bottom for simulation)

Bread Used	Meat Used	Cheese Used	Sandwiches	Excess Bread	Excess Meat	Excess Cheese
			Made			
5 slices	5 slices	5 slices				
4 slices	3 slices	2 slices				
			2 sandwiches	1 slice	0 slices	2 slices
6 slices			3 sandwiches		4 slices	1 slice

Part 2: Molecules:

<u>Make Water</u>

- 1. Now let's work with real chemical reaction, one that creates a very entertaining BOOM!
- 2. What is the mole ratio for the reaction of hydrogen and oxygen to produce water?

$_H_2 + _O_2 \rightarrow _H_2O$

3. Complete the table below while making water H_2O from hydrogen H_2 and oxygen O_2 :

Hydrogen Molecules H ₂	Oxygen Molecules O ₂	Water Molecules H ₂ O	Excess H ₂	Excess O ₂
4 molecules	4 molecules			
7 molecules	6 molecules			
		4 molecules	0 molecules	0 molecules
9 moles	8 moles			
		4 moles	1 moles	0 moles
4.0 moles	2.5 moles			
1.5 moles		1.5 moles	0 moles	0 moles

- 4. Notice that the labels changed from **molecules** to **moles**. This does not change the mole ratio, as a mole is simply a large number of molecules. How many molecules is a mole?
- 5. Now try producing **ammonia**, a very important chemical in industry and farming.
- 6. What is the mole ratio for the production of ammonia? $N_2 + H_2 \rightarrow NH_3$
- 7. Complete the table below:

Moles N ₂	Moles H ₂	Moles NH ₃	Excess N ₂	Excess H ₂
3 moles	6 moles			
6 moles	3 moles			
		4 moles	2 moles	0 moles
1.5 moles	4.0 moles			

- 8. Combustion of hydrocarbons like methane CH₄ produce two products, water and carbon dioxide CO₂.
- 9. What is the mole ratio for the combustion of methane? $_CH_4 + _O_2 \rightarrow _CO_2 + _H_2O$

10. Complete the table below: WATCH FOR FRACTIONS

mol CH ₄	mol O ₂	mol CO ₂	mol H ₂ O	Excess mol CH ₄	Excess mol O ₂
4 mol	4 mol				
3 mol	6 mol				
		2 mol	4 mol		
		3 mol		0 mol	
				2 mol	0 mol
				3 mol	1 mol



Part 3: Game:

Game: Level 1

Fill in the chart and include the correct formulas, no leftovers. Play for time after first time.

	Reactants	\rightarrow	Products	Best time
#1				
#2				
#3				
#4				
#5				

Game: Level 2

Fill in the chart and include the correct formulas, no leftovers. Play for time after first time.

	Reactants	\rightarrow	Products	Best time
#1				
#2				
#3				-
#4				-
#5				

Game: Level 3

Fill in the chart and include the correct formulas, no leftovers. Play for time after first time.

	Reactants	\rightarrow	Products	Best time
#1				
#2				
#3				
#4				
#5				